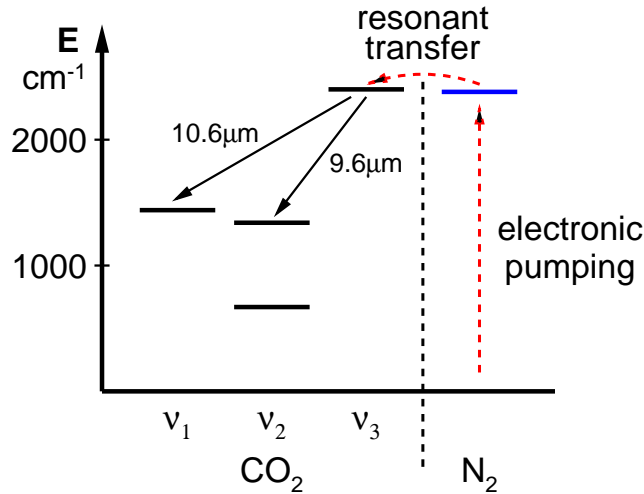


## 5 The CO<sub>2</sub> laser

The CO<sub>2</sub> laser works in the mid-IR range, 9–11  $\mu\text{m}$ , on ca. 100 vibration–rotation transitions. In its classical version it operates at a pressure of 10–20 mbar with a mixture of CO<sub>2</sub>:N<sub>2</sub>:He  $\simeq$  15%:15%:70%. The laser is pumped with current discharge, through a tube of 1–2 m, requiring voltages of 10–20 kV and a current of 10–20 mA. Glass is opaque at 10  $\mu\text{m}$  so transmissive optics are made of NaCl, ZnSe, Ge or other IR materials.



The vibration level structure of CO<sub>2</sub> and N<sub>2</sub> is shown in the figure. The  $v_1$  mode in CO<sub>2</sub> is the symmetric stretch vibration (the C atom stationary), the  $v_2$  mode is the bending vibration of the otherwise linear molecule and the  $v_3$  mode is the asymmetric stretch vibration.

The N<sub>2</sub> molecule provides an effective pump channel with near resonance with the upper laser level. The energy difference is only 18  $\text{cm}^{-1}$  compared to  $kT=208 \text{ cm}^{-1}$  at  $T = 300 \text{ K}$ . Helium acts as a cooling medium and speeds up the depopulation of the lower levels.

Each of the vibration levels is split into rotation levels marked with a rotation quantum number  $J$ . The rotational energy is given with

$$E_{\text{rot}} = BJ(J+1) \quad (1)$$

The energy constant  $B \simeq 0.387 \text{ cm}^{-1}$  is inversely proportional to the moment of inertia of the molecule. The  $J$  levels communicate fast with each other with a lifetime of only a few nanoseconds, whereas the lifetime of the vibration channels is tens of microseconds. We can therefore assume  $T_{\text{rot}} = T_{\text{transl}}$ . The rotational population distribution is given with  $N(J) = N_{\text{vibr}} f(J)$ , where

$$f(J) = [(2J+1) \exp(-BJ(J+1)/kT)]/Z_{\text{rot}} \quad (2)$$

$Z_{\text{rot}}$  is the partition function for the rotation levels.

Transitions are governed by the selection rules  $\Delta J = 0, \pm 1$ . Transitions with  $\Delta J = +1$  ( $J_{\text{lower}} - J_{\text{upper}}$ ) are called P-lines, lines with  $\Delta J = -1$  R-lines and lines with  $\Delta J = 0$  Q-lines. The CO<sub>2</sub> laser has only P- and R-lines since for symmetry reasons the upper laser level has only odd  $J$ -values and the lower levels only even  $J$ -values. The spectral lines are arranged in

bands Where the R-bands have higher frequency than the P-bands. There is a pair of bands for each of the lower vibration levels. At temperatures relevant to CO<sub>2</sub> laser discharge  $f_J$  peaks around  $J = 20$  and in each band lasing can be achieved at 20–25 lines. The total number of lines is thus 80–100 all in the range of 9–11  $\mu\text{m}$ .

Line separation is of the order 50 GHz while linewidths are  $\Delta\nu_D = 30 \text{ MHz}$  and  $\Delta\nu_L = 30 \text{ MHz}$ . Saturation intensity is of the order  $100 \text{ W/cm}^2$  which makes the CO<sub>2</sub> laser a powerful beast. The saturation intensity can be scaled up with pressure as  $p^2$ .